

Phytochemical Study and Antioxidative Property of Ethanolic Extract from *Termitomyces clypeatus*

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ABSTRACT

The present study documents the phytochemical screening and antioxidant properties of ethanolic extract of *Termitomyces clypeatus*. Phytochemical constituents like phenols, flavonoids and ascorbic acid were much higher than β carotene and lycopene. The analysis of the phenolic compounds performed by HPLC, revealed the presence of pyrogallol (0.223 $\mu\text{g}/\text{mg}$) and cinnamic acid (0.095 $\mu\text{g}/\text{mg}$). Antioxidant activity was evaluated through superoxide radical scavenging assay, DPPH radical scavenging assay, chelating ability, reducing power and total antioxidant capacity determining assays. Their EC_{50} values ranged from 0.21 $\mu\text{g}/\text{ml}$ to 3.22 mg/ml . Result implies that *T. clypeatus* can not only serve as a food supplement but also be used as treatment for various oxidative stress related diseases.

INTRODUCTION

Free radicals and degenerative diseases are associated with aging and include cancer, cardiovascular disease, immune-system decline, brain dysfunction and cataracts (Ames *et al.*, 1993; Khatua *et al.*, 2013). As oxidative damage to DNA, proteins and other macromolecules accumulates with age; leads to aging (Fraga *et al.*, 1990; Harman, 1981). Free radicals such as superoxide, hydrogen peroxide and hydroxyl radicals, which are mutagens produced by radiation, are also by-products of normal metabolism (Sies, 1986; Wagner *et al.*, 1992). Our own immune system at times is unable to combat these reactive oxygen species (ROS). Hence, the need for antioxidants creeps in. As carcinogenic properties have been reported for some synthetic antioxidants, research in the last three decades on the potential use of natural antioxidants from plants and mushrooms have received much importance. Extracts from mushroom have received attention based on their safety and records of health promotion. It has been also established that they are less toxic. They may act directly as antioxidant or prevent underlying

oxidative stress related pathological conditions such as cancer (Chatterjee *et al.*, 2014), heart ailments (Biswas *et al.*, 2011), diabetes (Biswas and Acharya, 2013), inflammation (Biswas *et al.*, 2010), gastric ulcer (Chatterjee *et al.*, 2013), hepatic damage (Acharya *et al.*, 2012; Chatterjee *et al.*, 2012), microbial pathogens (Rai *et al.*, 2013), parasitic organisms (Mallick *et al.*, 2014, Mallick *et al.*, 2015) etc. *Termitomyces clypeatus* R. Heim commonly known as “bali chatu” by the local people of lateritic zone of West Bengal has been traditionally used as food for their flavour, texture and delicacy. Recently a water soluble pure polysaccharide with average molecular weight of $\sim 1.98 \times 10^5$ Da was isolated from *T. clypeatus* having strong antioxidative properties (Pattanayak *et al.*, 2015). As far as our literature survey says the antioxidant activity of ethanolic rich fraction of this mushroom from West Bengal has not yet been published. In this scenario the present investigation was done. We have examined the antioxidant activity of ethanolic fraction of *T. clypeatus* employing various *in vitro* assay models such as superoxide radical scavenging activity, DPPH radical scavenging activity, chelating of ferrous ion, reducing ability and total antioxidant capacity along with a phytochemical screening for determining the usefulness of this mushroom as a functional food.

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MATERIALS AND METHODS

Collection, Preparation of sample and authentication

The mushroom *T. clypeatus* was collected from West Midnapore district of West Bengal. With proper scientific measures they were brought to the laboratory and cleaned well. Identification was done following standard literature (Karun and Sridhar, 2013). The voucher specimen has been deposited in the Calcutta University Herbarium (CUH) with the accession number CUH AM 351.

Ethanolic fraction was extracted according to the method of Dasgupta *et al.* (2013). Dried and powdered basidiocarps of *T. clypeatus* were extracted with ethanol at 25°C for 2 days and filtered. After filtration, the residue was re-extracted with ethanol. After filtration the filtrate was concentrated under reduced pressure in a rotary evaporator. Now, this concentrated ethanolic extract of *T. clypeatus* was stored at 4°C. The percentage yield extracts were calculated based on dry weight as:

$$\text{Yield (\%)} = (W_1 \times 100) / W_2$$

Where W_1 = weight of extract after solvent evaporation; W_2 = Weight of the minced mushroom.

Phytochemical screening

The content of total phenolic compounds in the ethanolic extract of *T. clypeatus* was estimated using Folin-Ciocalteu reagent (Singleton and Rossi 1965). Gallic acid (10-40 µg) was used as a standard and expressed as µg/gallic acid equivalents/g of extract. Aluminium nitrate and potassium acetate were required to determine total flavonoid content (Park *et al.*, 1997) and quercetin (5–20 µg) was used as a standard and presented as µg of quercetin equivalents/ g of extract. Quantification of ascorbic acid was done by titration against 2, 6-dichlorophenol indophenol dye using oxalic acid (Rekha *et al.*, 2012). Beta-carotene and lycopene were estimated by measuring absorbance at 453, 505 and 663 nm (Nagata and Yamashita, 1992).

Detection of Phenols and flavonoids by HPLC

The phenolic profile of the extract was determined using eleven standards of Sigma Aldrich (MO, USA) like gallic acid, chlorogenic acid, vanillic acid, p-coumaric acid, ferulic acid, myricetin, salicylic acid, quercetin, cinnamic acid, pyrogallol and kaempferol. For quantitative analysis, a calibration curve (10–50 µg/ml) for each phenolic standard was constructed.

Antioxidant Activity

Following methods of Martinez *et al.* (2001) superoxide radical scavenging activity of the ethanolic extract of *T. clypeatus* (0.2 mg/ml-0.6 mg/ml) was determined. BHA was used as a positive control. DPPH radical scavenging activity was determined for concentrations 2.0 mg/ml to 3.0 mg/ml following the protocol of Shimada *et al.* (1992). Various concentrations (0.1-0.3 mg/ml) of the extract were checked if they can chelate ferrous ion, based on methods stated by Dinis *et al.* (1994). Reducing power of the extract was determined spectrophotometrically at 700 nm as per

methods of Oyaizu (1986) using ascorbic acid as standard versus different concentrations (1.0-2.0 mg/ml). Total antioxidant capacity assay was also carried out as described by Prieto *et al.* (1999) with little modification (Mitra *et al.*, 2014). The activity was expressed as the number of equivalents of ascorbic acid (AAE). EC_{50} value is the half maximal effective value i.e. the concentration of extract providing 50% of antioxidant activity or 0.5 absorbance. Graphs were plotted based on above data and respective EC_{50} values were determined from them.

Statistical analysis

Data were expressed as mean \pm S.D. (Standard deviation). Means of triplicate analyses were calculated. The Student's t test was used for comparison between standard and the sample. A difference was considered to be statistically significant when $p < 0.05$.

RESULTS AND DISCUSSION

The extractive yield of the ethanolic extract of *T. clypeatus* was 7%. It was seen to have a good number of phytochemicals, each of which was responsible to give it a good antioxidative property. Phenolic compounds are known to be powerful chain-breaking antioxidants. Their hydroxyl groups provide them scavenging ability. In this study, the total phenolic content of the extract was noted to be 5 ± 0.48 µg/mg. Its estimated flavonoid content was 2.14 ± 0.47 µg/mg. Ascorbic acid was reported to interact directly with radicals such as O_2^- and OH in plasma, thus preventing damage of cells. Here, ascorbic acid content was seen to be 1.3 ± 0.11 µg/mg. β-carotene protects against cancer and cardiovascular diseases. Lycopene, on the other hand is one of the antioxidants that prevents carcinogenesis and atherogenesis. β-carotene and lycopene were found in very less amounts in this extract viz., 0.0014 ± 0.0002 and 0.0011 ± 0.0002 µg/mg respectively.

HPLC has been done to predict phenolic composition of a fraction extracted after thermal processing. As shown in figure 1a and 1b, eleven phenolic substances were analysed and two of them were detected in the ethanolic extract of *T. clypeatus*. Our findings revealed that the dominant phenolic compounds were pyrogallol (0.223 µg/mg) and cinnamic acid (0.095 µg/mg). The response of antioxidants to different radical or oxidant sources may be different. Therefore, no single assay can be capable to reflect the mechanism of action of all radical sources or antioxidants in any system. This way, the antioxidant activity of the samples was assessed through five different methods. Superoxide anion, which is a reduced form of molecular oxygen, has been implicated in initiating oxidation reactions associated with aging (Lavhale and Mishra, 2007). With one unpaired electron, the superoxide ion is a free radical, and, like di-oxygen, it is paramagnetic. It plays an important role in formation of other reactive oxygen species, which induce oxidative damage in lipids, proteins and DNA. The extract showed superoxide radical scavenging capacity in a concentration dependant manner (Figure 2 a). The EC_{50} value was 0.33 ± 0.01 mg/ml.

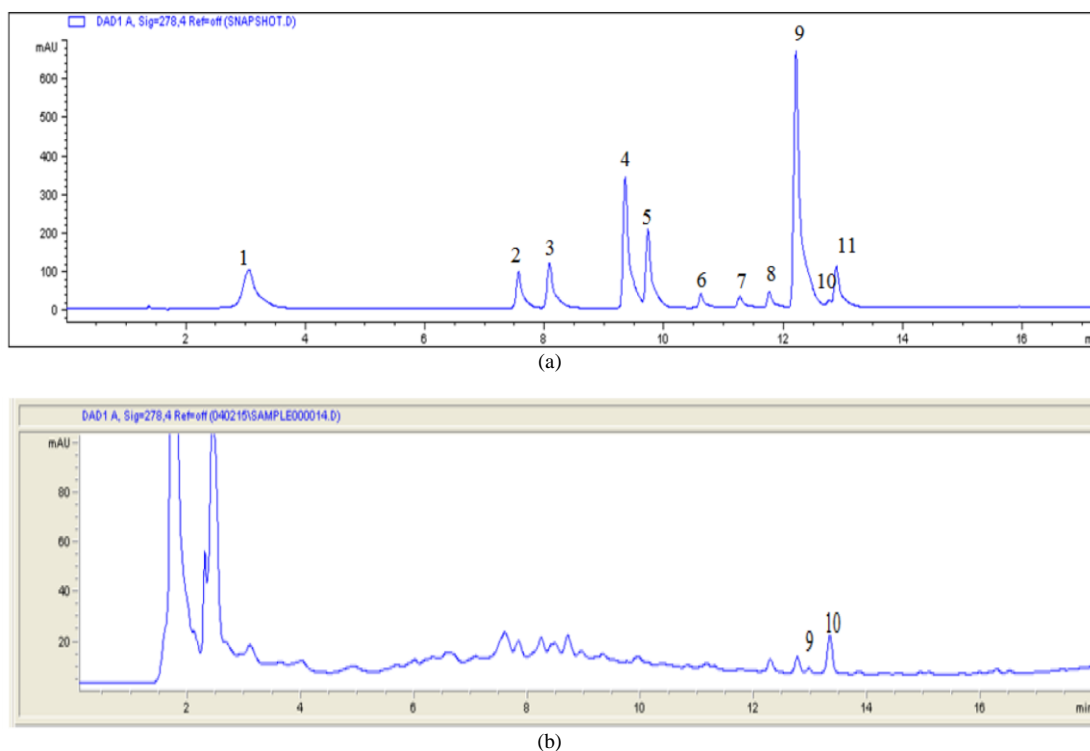


Fig. 1: HPLC chromatogram. (a) Peaks of standards: 1: gallic acid, 2: chlorogenic acid, 3: vanillic acid, 4: *p*-coumaric acid, 5: ferulic acid, 6: myricetin, 7: salicylic acid, 8: quercetin, 9: cinnamic acid, 10: pyrogallol, 11: kaempferol. (b) Peaks of ethanolic extract of *Termitomyces clypeatus*: (9) cinnamic acid, (10) pyrogallol.

The method is based on the auto oxidation of riboflavin in presence of light which in turn reduced NBT to form a blue colour formazan. EC_{50} value of it was much lower than that of ethanolic extract of *Termitomyces medius* (1.40 mg/ml) and *Russula albonigra* (0.74 mg/ml) (Mitra *et al.*, 2014; Dasgupta *et al.*, 2014).

The DPPH radical is a stable radical with a maximum absorbance at 517 nm that can readily undergo reduction by any antioxidant. The ease and convenience of this reaction has resulted in the widespread use of it in the free radical-scavenging activity assessment (Kumar *et al.*, 2008). The DPPH radical scavenging activity of the ethanolic extract is shown in Figure 2 b. and calculated EC_{50} is only at a concentration of 3.22 ± 0.4 mg/ml.

Iron generates free radical by the Fenton & Haber-Weiss reaction. Chelation of metal ions prevents oxyradical generation and consequent oxidative damage. Chelation is an important antioxidant mechanism because it reduces concentration of the catalysing transition metal in LPO (Kumar *et al.*, 2008). The ethanolic extract has demonstrated reasonable ferrous ion chelating efficacy (Figure 2 c.) with EC_{50} value at 0.21 ± 0.03 mg/ml, which indicated that it was a far better chelator than *Termitomyces medius* (0.68 mg/ml), *Russula albonigra* (0.81 mg/ml) and *Amanita vaginata* (0.73 mg/ml) (Mitra *et al.*, 2014, Dasgupta *et al.*, 2014, Paloi and Acharya, 2014). It is also reported that chelating agents that form σ -bonds with a metal, are effective as secondary antioxidants as they reduce the redox potential,

thereby stabilizing the oxidised form of the metal (Srivastava *et al.*, 2006). The reducing power assay is based on the ability of sample to reduce yellow ferric form to blue ferrous form by the action of electron-donating antioxidants (Benzie and Szeto, 1999). The extract could reduce Fe^{3+} to Fe^{2+} with 50 % inhibition capacity at a concentration of 1.77 ± 0.035 mg/ml (Figure 2 d.). In comparison to this the EC_{50} value of ethanolic extract of *Termitomyces medius* was much higher i.e. 2.05 mg/ml (Mitra *et al.*, 2014 a). Reductones work by breaking the chain of free radicals by donating hydrogen atoms (Mitra *et al.*, 2014 b). This change can be monitored at 700nm, by measuring the intensity of the Perl's Prussian blue colour. In the designed experiments the extract's reducing power was compared to that of BHA, a synthetic antioxidant.

Total antioxidant capacity on the other hand was measured by the formation of green phosphomolybdenum complex. The ethanolic extract resulted in the reduction of Mo (VI) to Mo (V) and form a green phosphate/Mo (V) complex. The colour intensity was determined with the maximal absorption at 695 nm. Ascorbic acid was used as standard. Total antioxidant capacity measured had shown that 1mg of extract was as functional as 1.3 ± 0.39 μ g of ascorbic acid (expressed as 100 μ g AAE).

There were statistically significant differences between the ethanolic extract of *T. clypeatus* and the respective standards ($p < 0.05$).

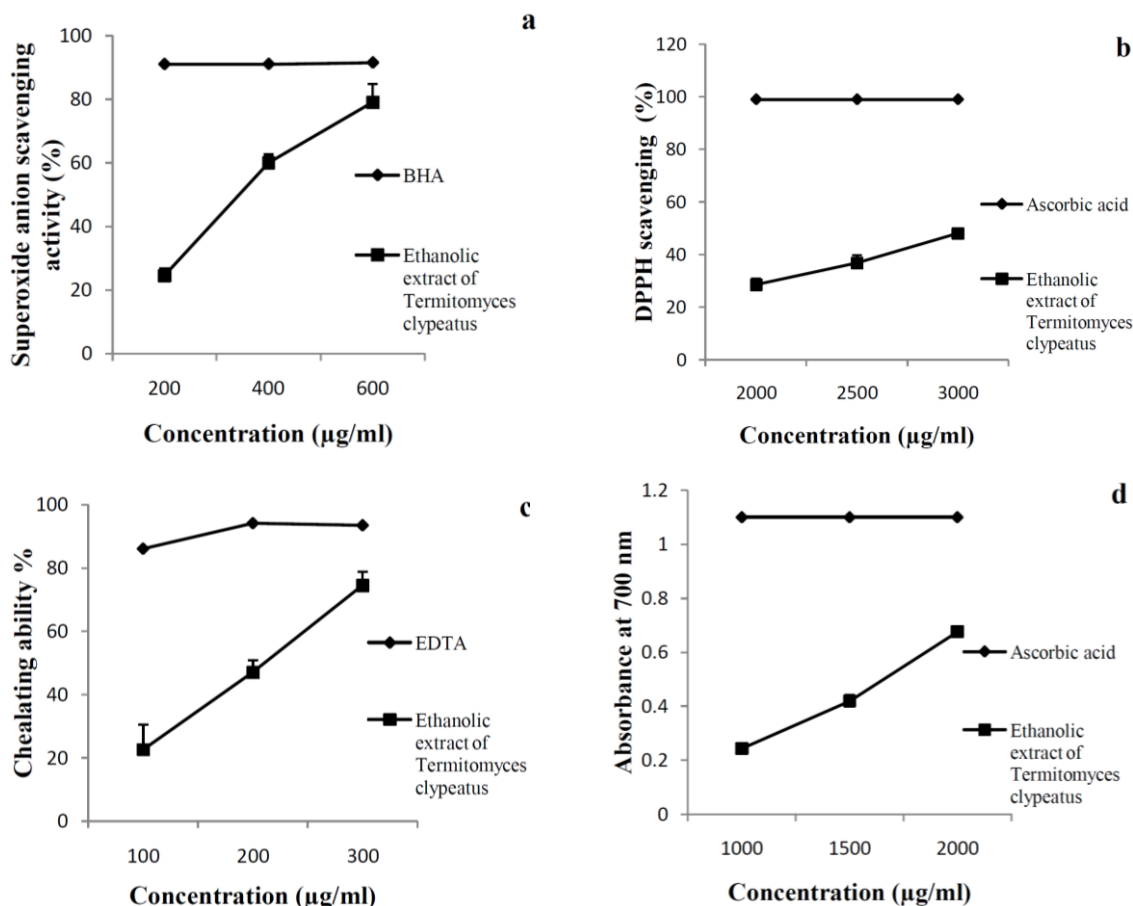


Fig. 2: Antioxidative activities of ethanolic fraction of *Termitomyces clypeatus*: (a) Superoxide radical scavenging activity (b) DPPH radical scavenging activity (c) Ferrous ion chelating ability and (d) Reducing power. Results are the mean \pm SD of three separate experiments, each in triplicate.

CONCLUSION

Overall, it has been verified that wild ethanolic extract of edible mushroom *T. clypeatus* has a variety of phytochemicals such as phenols, ascorbic acid, flavonoids, β carotene and lycopene which contribute to the good antioxidant mechanisms viz., superoxide, DPPH radical scavenging activity, chelation of ferrous ion activity and reducing ability. Therefore, food modification through the balanced consumption of this mushroom is more effective and important than other nutritional supplements for the primary prevention of acute diseases.

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