

Electronic Supplementary Information

Fast identification and quantification of BTEX by Raman spectrometry and chemometrics

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Tables S-1 and S-2 exhibit the actual concentrations of each BTEX in all the samples for Raman measurements. Samples 55 to 72 were used to determine spectral uncertainty and to check sample preparation uncertainty.

Figure S-1 shows the comparison between the previous concentrations and those calculated by the data processing (SIMPLISMA 1 way from chart 1 in the article) for each BTEX.

Figure S-2 displays a 3D bar graph representing the concentrations of the last samples (table S-2) obtained thanks to the data processing.

Table S-1. Experiment plan, part 1. Concentration values in g/L ($\pm 5\%$).

S. n°	B.	T.	E.	o-X.	m-X.	p-X.	S. n°	B.	T.	E.	o-X.	m-X.	p-X.
01	153	142	143	147	143	141	26	14	0	0	541	264	56
02	287	19	38	77	151	297	27	439	430.3	0	0	0	0
03	288	299	19	39	75	148	28	438	0	432	0	0	0
04	287	149	301	19	38	74	29	432	0	0	444	0	0
05	283	75	150	309	19	37	30	438	0	0	0	433	0
06	78	75	601	39	38	37	31	441	0	0	0	0	426
07	24	599	75	19	38	111	32	863	11	0	0	0	0
08	287	150	301	19	38	74	33	863	0	11	0	0	0
09	794	22	45	0	0	11	34	863	0	0	11	0	0
10	370	0	376	39	11	74	35	863	0	0	0	11	0
11	422	11	0	386	19	37	36	836	0	0	0	0	11
12	115	150	150	155	151	148	37	840	0	0	11	11	11.2
13	836	7.4	7.4	7.4	7.4	7.3	38	433	0	0	386	38	19
14	829	38	7.4	0	0	0	39	8.3	0	214	220	215	211
15	710	0	0	116	11	37	40	188	0	601	11	23	44
16	142	374	188	97	45	22	41	188	0	601	46	11	22
17	1.7	172	173	178	173	171	42	65	0	601	155	38	11
18	10	299	368	39	113	37	43	4.8	0	11	711	76	74
19	1.7	0	38	232	262	371	44	9.6	0	19	97	56	682
20	14	0	0	541	264	56	45	115	150	150	155	151	148
21	115	150	150	155	151	148	46	874	0	0	0	0	0
22	115	150	150	155	151	148	47	0	865	0	0	0	0
23	115	150	150	155	151	148	48	0	0	867	0	0	0
24	115	150	150	155	151	148	49	0	0	0	879	0	0
25	115	150	150	155	151	148	50	0	0	0	0	868	0
							51	0	0	0	0	0	861

S. n°: Sample n°, B.: Benzene, T.: Toluene, E.: Ethylbenzene, o-X.: ortho-Xylene, m-X.: meta-Xylene, p-X.: para-Xylene, Chlf: Chloroform.

Table S-2. Experiment plan, part 2. Concentration values in g/L ($\pm 5\%$).

S. n°	B.	T.	E.	o-X.	m-X.	p-X.	Chlf	S. n°	B.	T.	E.	o-X.	m-X.	p-X.
51b	0	0	0	0	0	0	$1.49 \cdot 10^6$	63	288	19	38	77	151	297
52	393	433	41	0	0	2.2	0	64	1.7	0	38	232	262	371
53	153	151	108	0	22	431	0	65	1.7	0	38	232	262	371
54	437	0	0	308	109	22	0	66	1.7	0	38	232	262	371
51c	140	138	139	141	139	137	60	67	1.7	0	38	232	262	371
51d	109	108	108	110	109	108	393	68	1.7	0	38	232	262	371
55	288	19	38	77	151	297	0	69	1.7	0	38	232	262	371
56	288	19	38	77	151	297	0	70	1.7	0	38	232	262	371
57	288	19	38	77	151	297	0	71	1.7	0	38	232	262	371
58	288	19	38	77	151	297	0	72	1.7	0	38	232	262	371
59	288	19	38	77	151	297	0							
60	288	19	38	77	151	297	0							
61	288	19	38	77	151	297	0							
62	288	19	38	77	151	297	0							

S. n°: Sample n°, B.: Benzene, T.: Toluene, E.: Ethylbenzene, o-X.: ortho-Xylene, m-X.: meta-Xylene, p-X.: para-Xylene, Chlf: Chloroform.

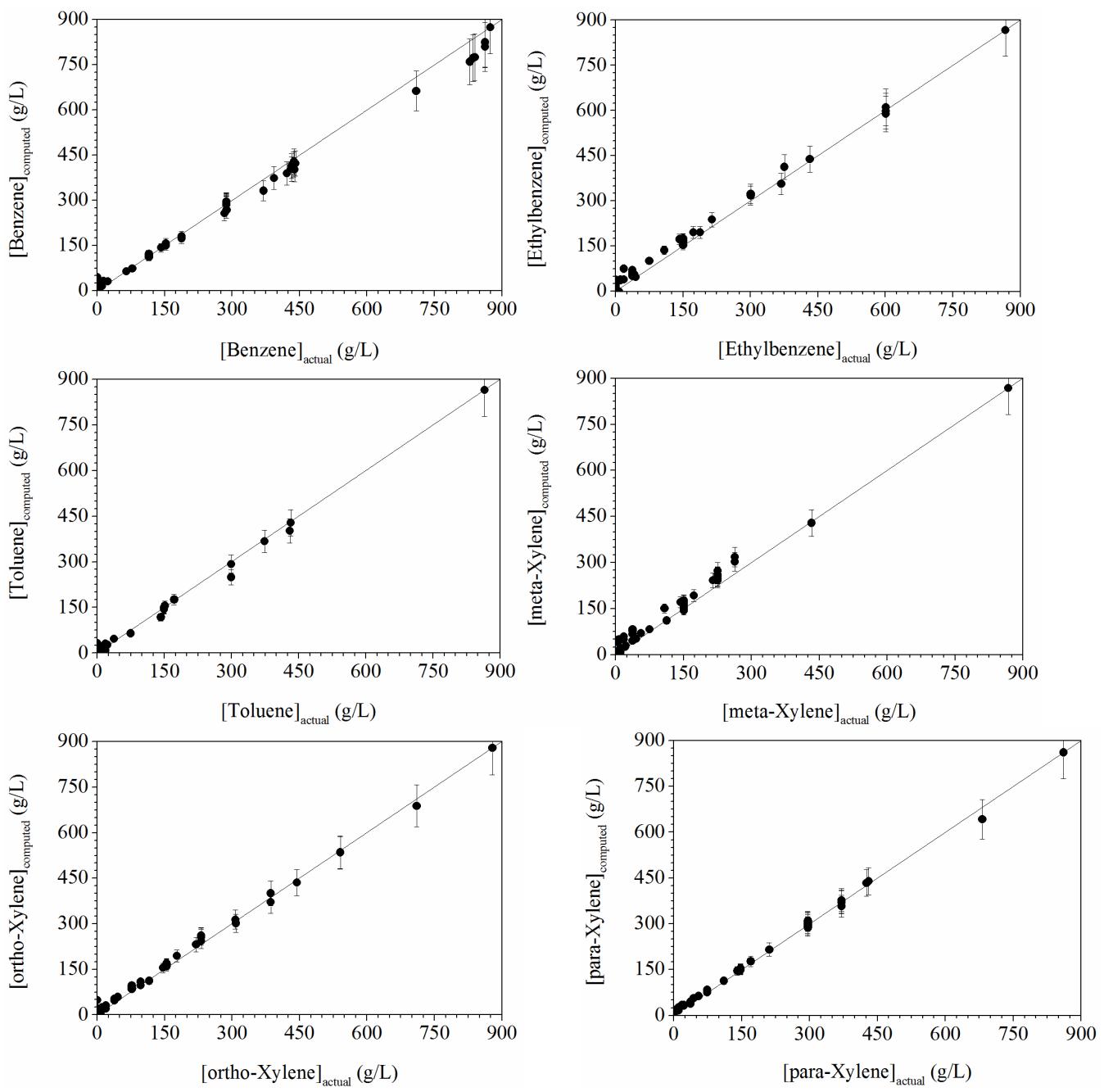


Figure S-2. Plots of computed vs actual concentrations of BTEX compounds in different mixtures using resolved spectra calculated by SIMPLISMA.

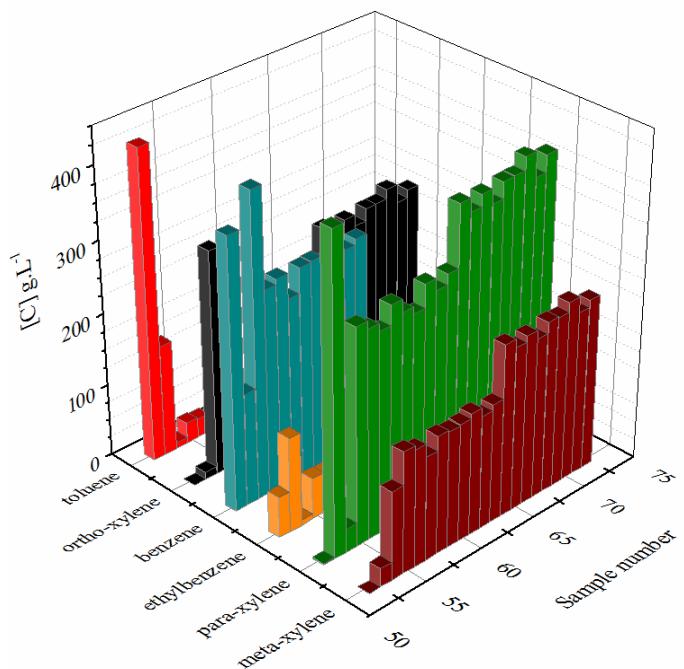


Figure S-3. Graphical results of the data processing showing the concentration of each BTEX compound from sample 52 to 72.