

*Supplementary Information for:*

**Mechanisms insight into oxygen reduction reaction on the dual FeN<sub>2</sub> embedded graphene for  
proton exchange membrane fuel cells**

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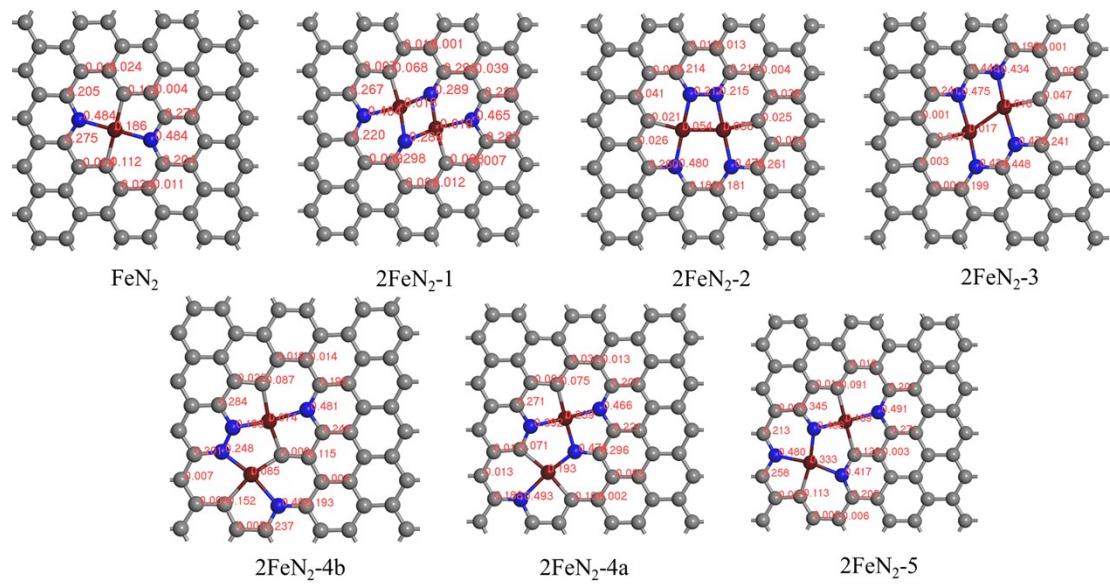


Fig. S1. Mulliken charges of these catalysts.

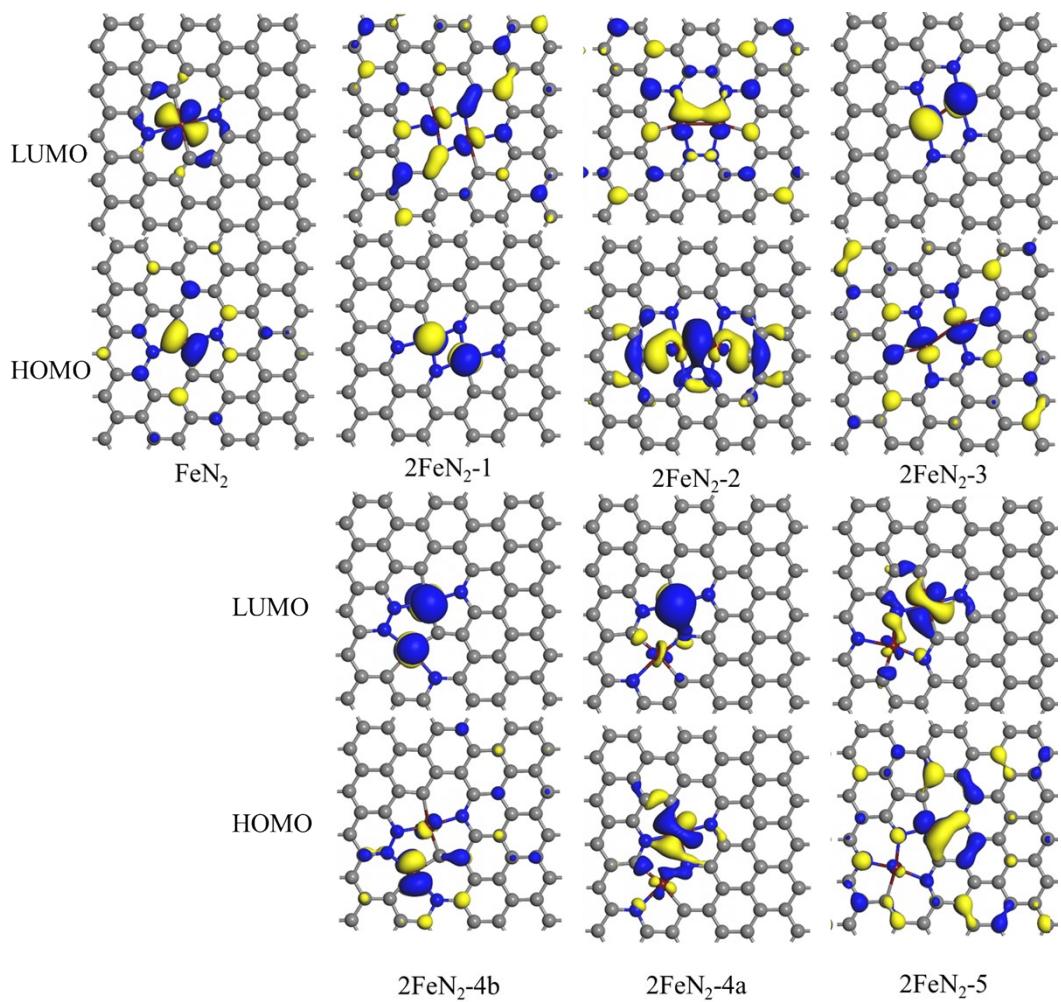


Fig. S2. HOMO and LUMO of these catalysts.

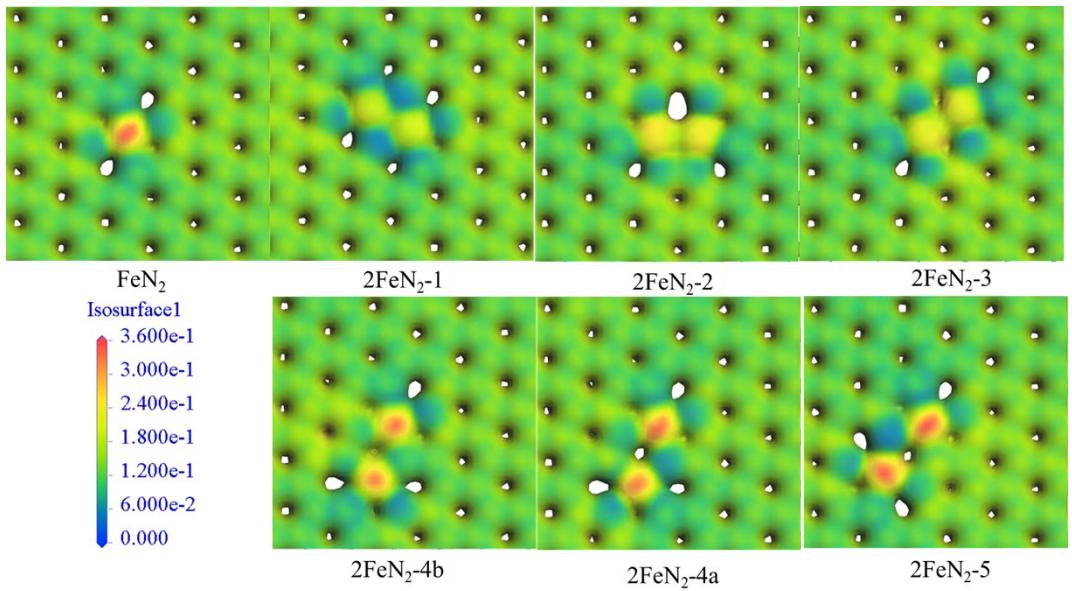


Fig. S3. the electrostatic potential surface of these catalysts.

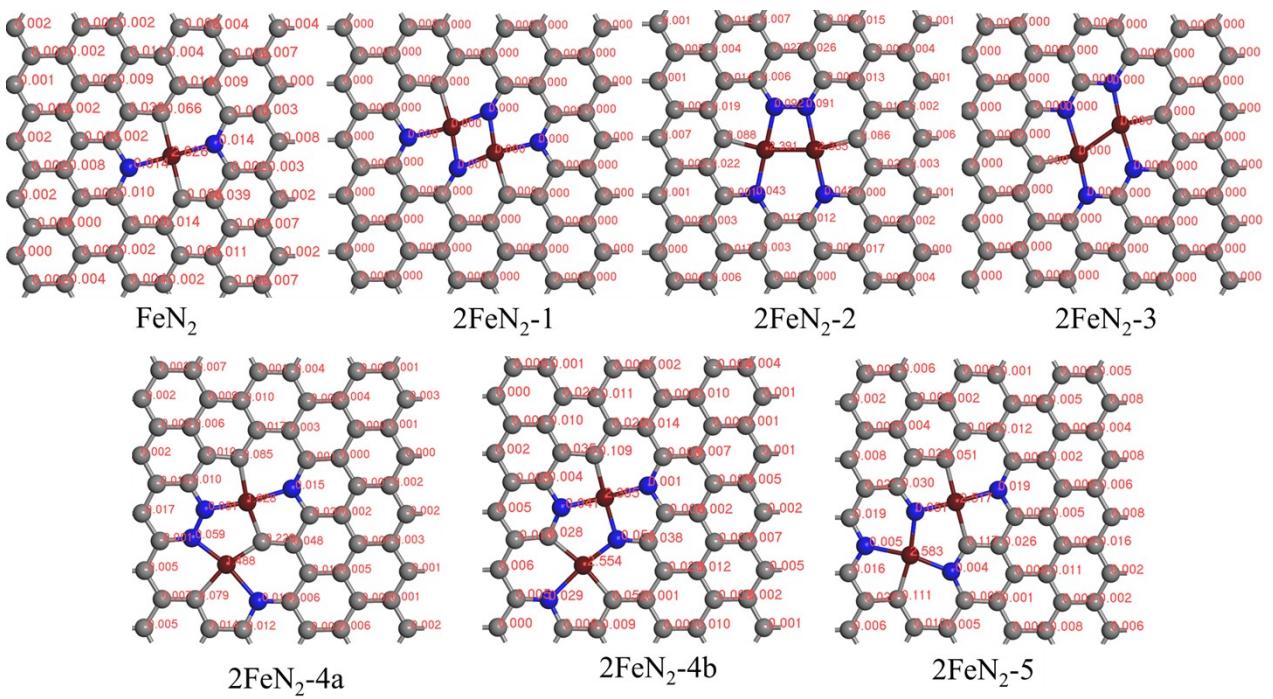


Fig. S4. The magnetic moments on Fe atom of these catalysts.

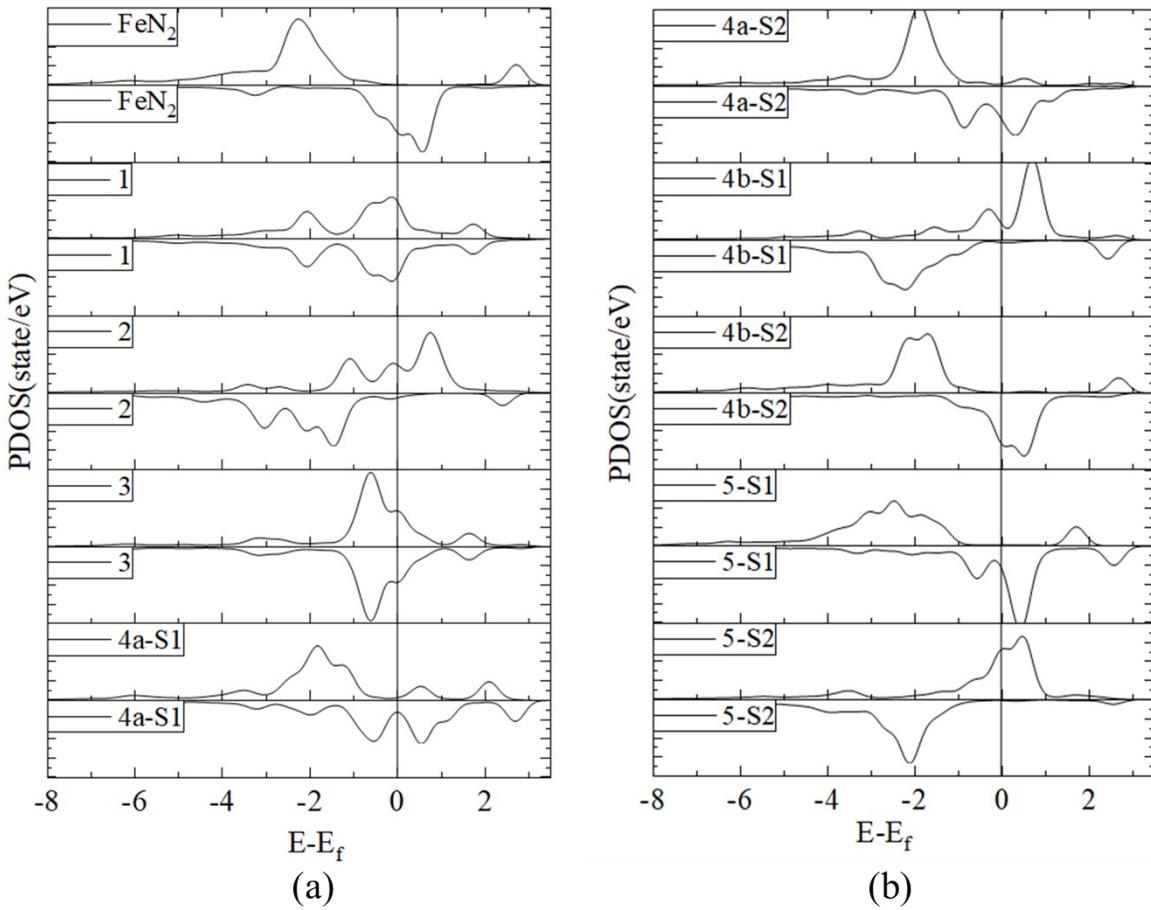


Fig.S5. PDOS for the Fe atom of these catalysts, the value of the PDOS varies in a range of -3.3 eV to 3.3 eV.

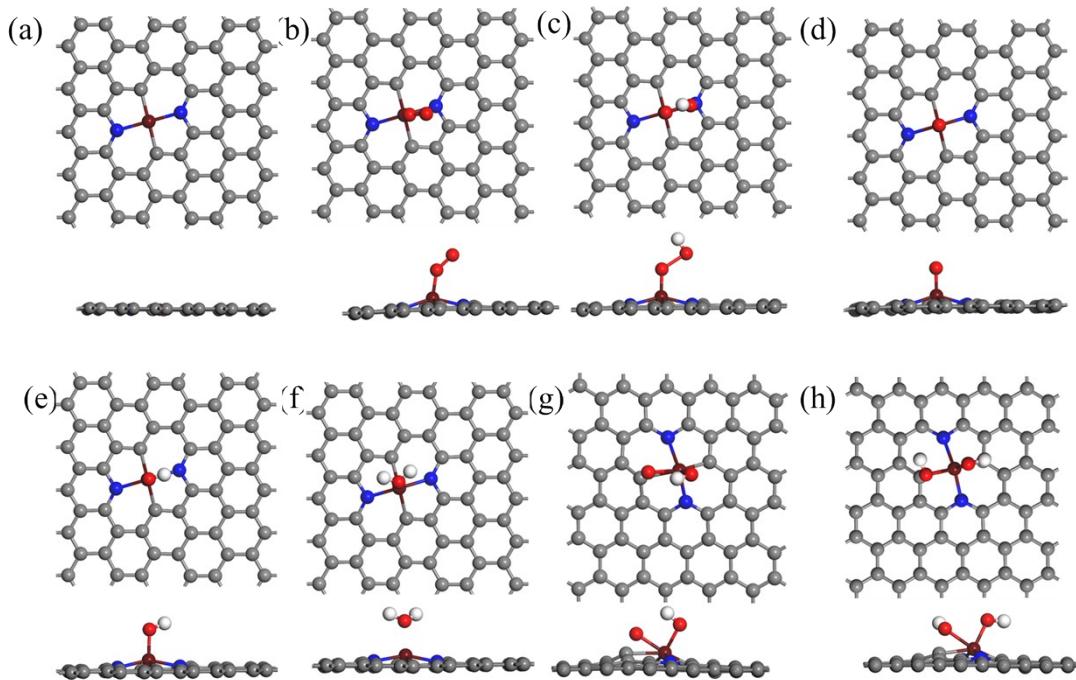


Fig. S6 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{FeN}_2/\text{G}$  catalytic surface. (a)\*, (b)  
\* $\text{O}_2$ , (c) \* $\text{OOH}$ , (d) \* $\text{O}$ , (e) \* $\text{OH}$ , (f) $\text{H}_2\text{O}^*$ , (g) \* $\text{O}^+$ \* $\text{OH}$ , (h) \* $\text{OH}^+$ \* $\text{OH}$ .

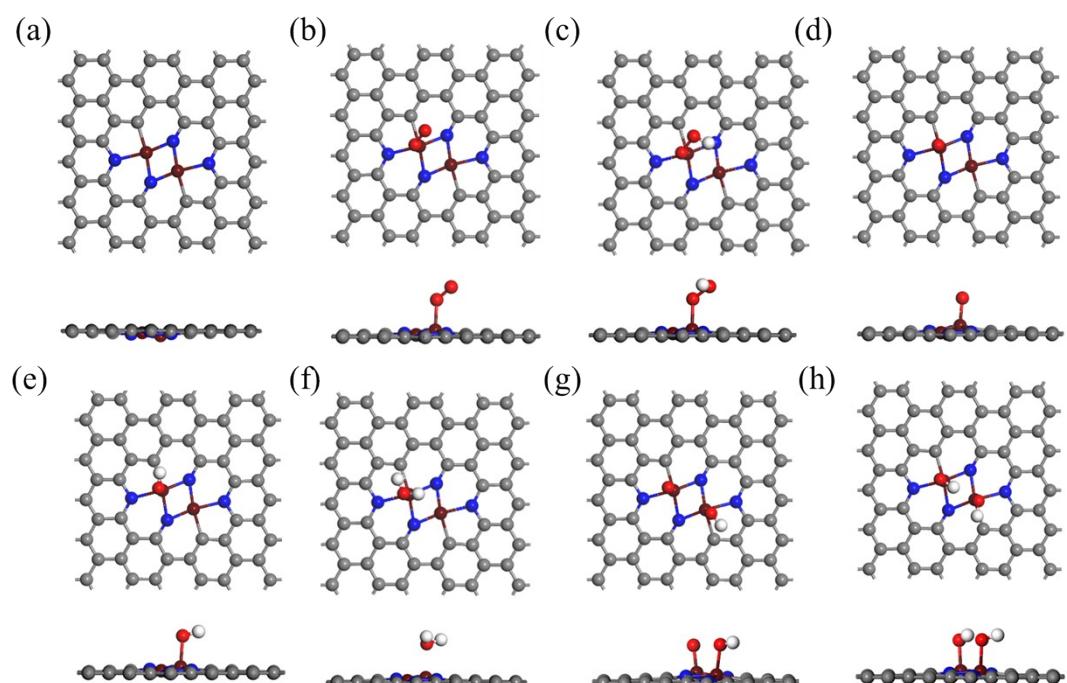


Fig. S7 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4-1/\text{G}$  catalytic surface. (a)\*, (b)  
\* $\text{O}_2$ , (c) \* $\text{OOH}$ , (d) \* $\text{O}$ , (e) \* $\text{OH}$ , (f) $\text{H}_2\text{O}^*$ , (g) \* $\text{O}^+$ \* $\text{OH}$ , (h) \* $\text{OH}^+$ \* $\text{OH}$ .

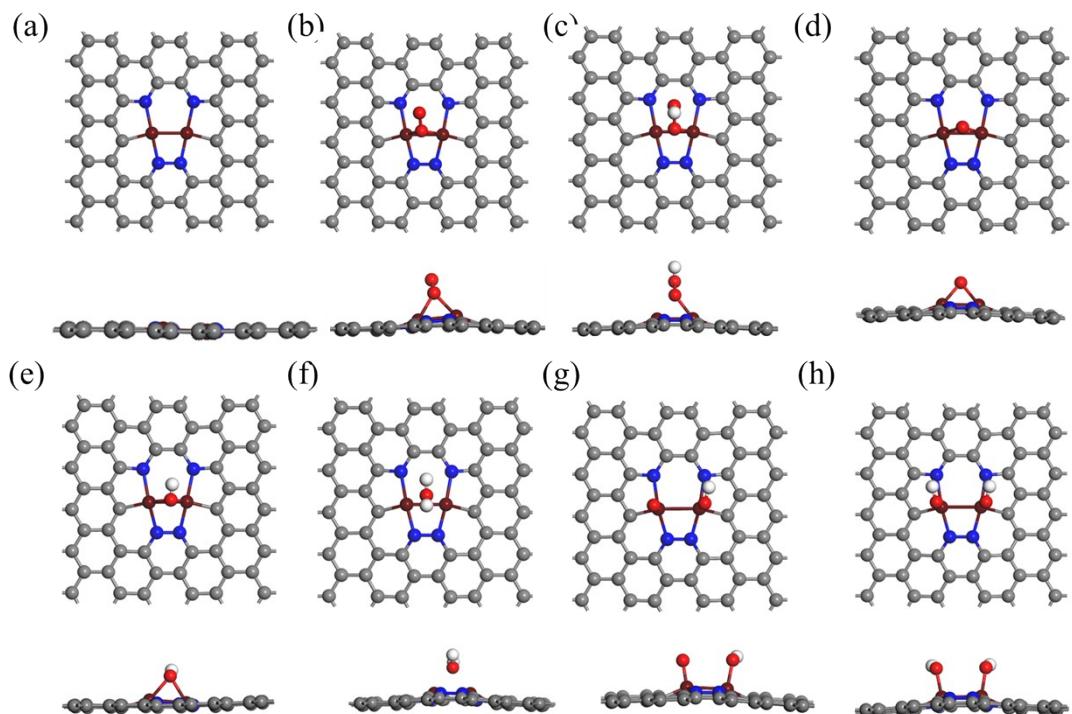


Fig. S8 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-}2/\text{G}$  catalytic surface. (a)\*, (b)  
\* $\text{O}_2$ , (c) \* $\text{OOH}$ , (d) \* $\text{O}$ , (e) \* $\text{OH}$ , (f) $\text{H}_2\text{O}^*$ , (g) \* $\text{O}^*+\text{OH}$ , (h) \* $\text{OH}^*+\text{OH}$ .

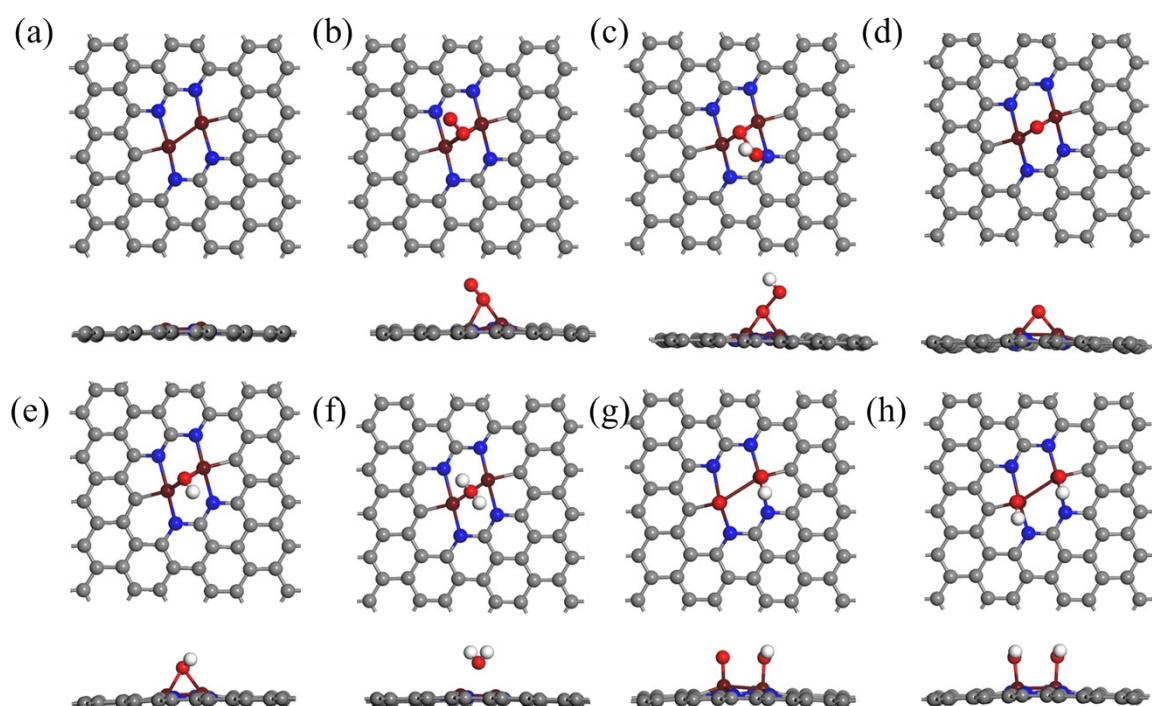


Fig. S9 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-}3/\text{G}$  catalytic surface. (a)\*, (b)  
\* $\text{O}_2$ , (c) \* $\text{OOH}$ , (d) \* $\text{O}$ , (e) \* $\text{OH}$ , (f) $\text{H}_2\text{O}^*$ , (g) \* $\text{O}^*+\text{OH}$ , (h) \* $\text{OH}^*+\text{OH}$ .

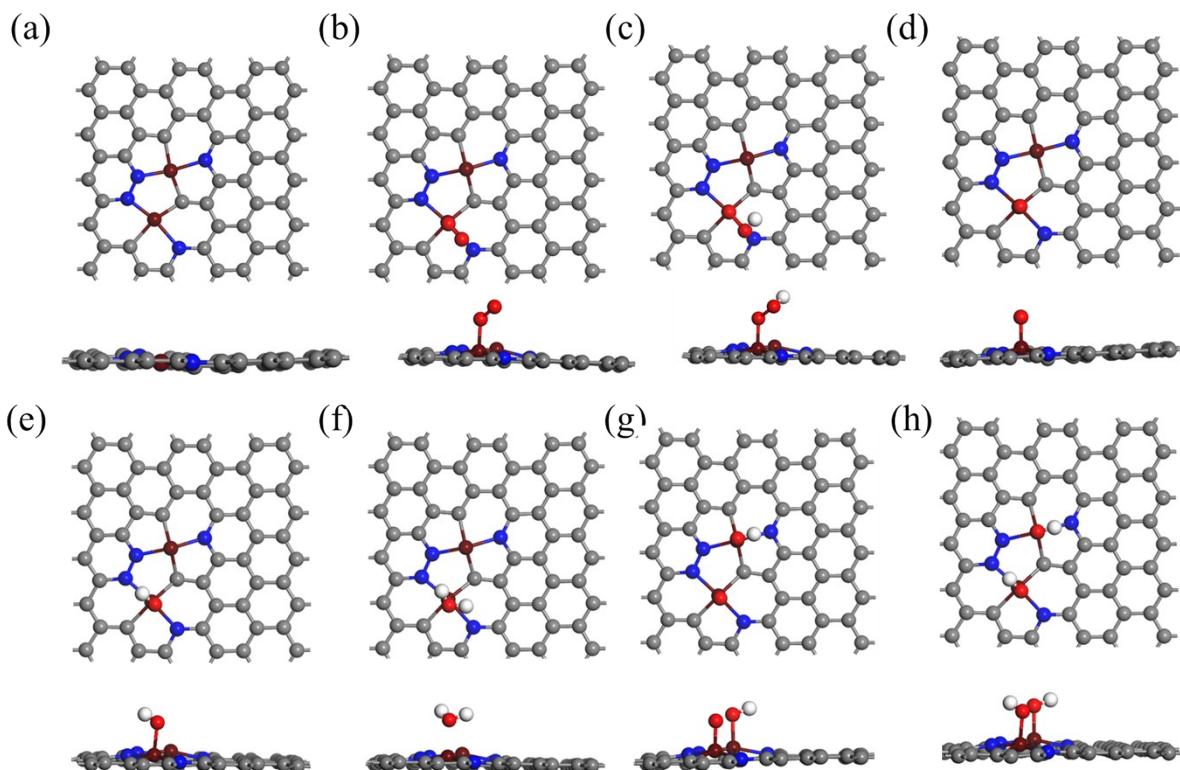


Fig. S10 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-}4\text{a}\text{-S1}/\text{G}$  catalytic surface.  
 (a)\*, (b)  $^*\text{O}_2$ , (c)  $^*\text{OOH}$ , (d)  $^*\text{O}$ , (e)  $^*\text{OH}$ , (f)  $\text{H}_2\text{O}^*$ , (g)  $^*\text{O}^+\text{*OH}$ , (h)  $^*\text{OH}^+\text{*OH}$ .

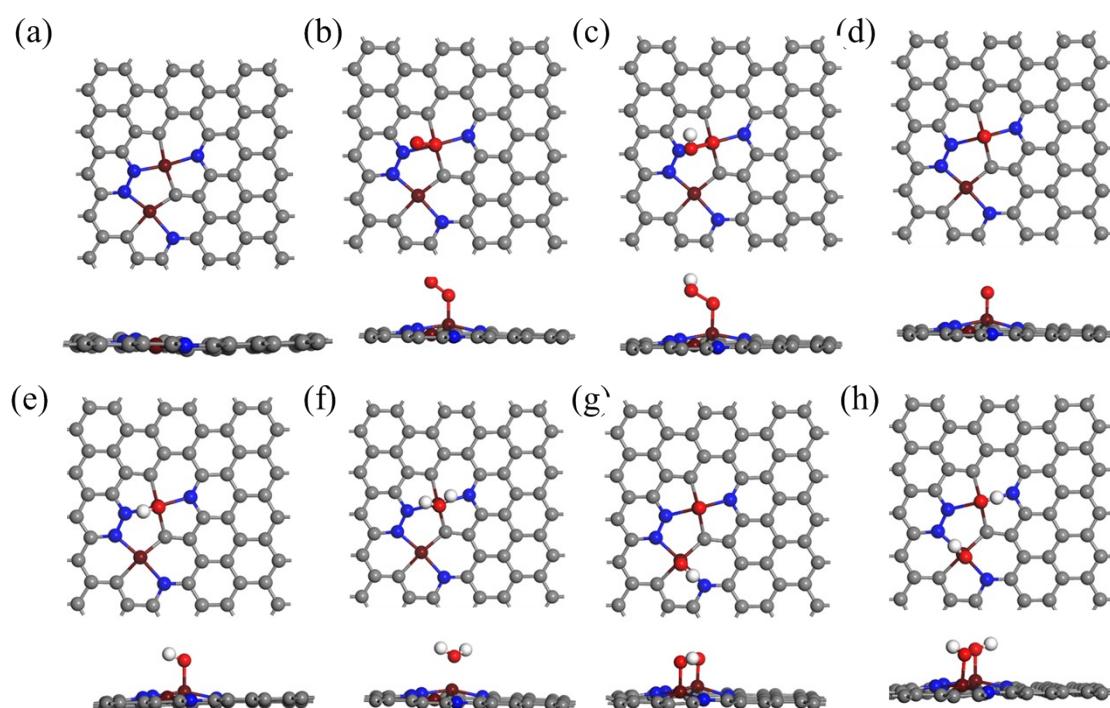


Fig. S11 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-}4\text{a}\text{-S2}/\text{G}$  catalytic surface.  
 (a)\*, (b)  $^*\text{O}_2$ , (c)  $^*\text{OOH}$ , (d)  $^*\text{O}$ , (e)  $^*\text{OH}$ , (f)  $\text{H}_2\text{O}^*$ , (g)  $^*\text{O}^+\text{*OH}$ , (h)  $^*\text{OH}^+\text{*OH}$ .

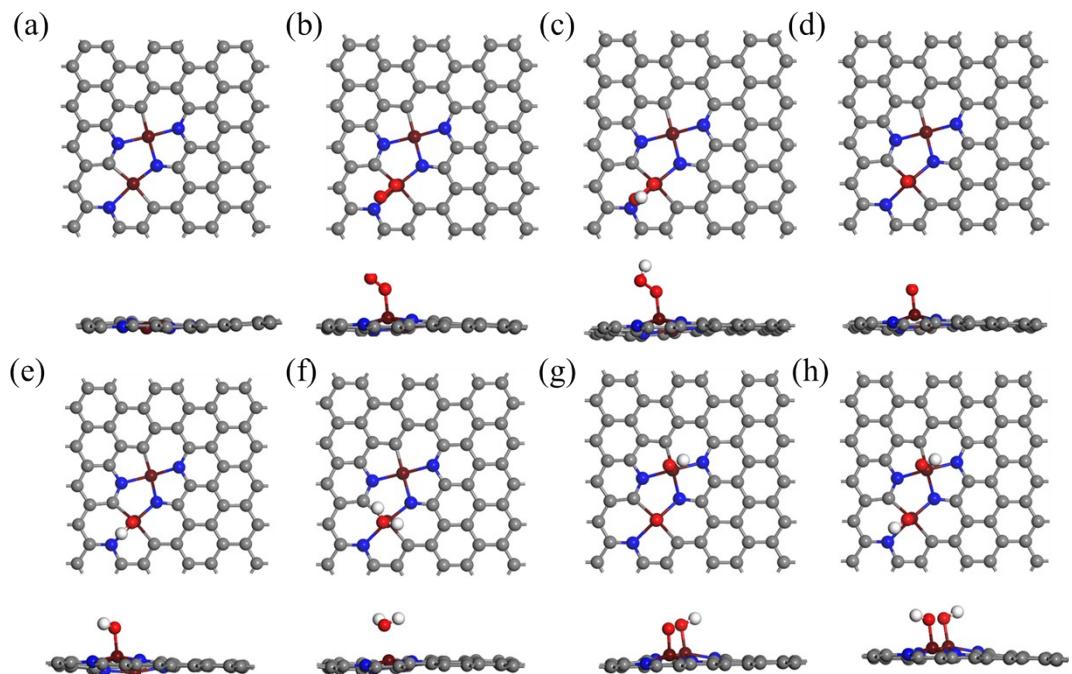


Fig. S12 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-}4\text{b}\text{-S1}/\text{G}$  catalytic surface.  
(a)\*, (b)  $^*\text{O}_2$ , (c)  $^*\text{OOH}$ , (d)  $^*\text{O}$ , (e)  $^*\text{OH}$ , (f)  $\text{H}_2\text{O}^*$ , (g)  $^*\text{O}^+*\text{OH}$ , (h)  $^*\text{OH}^+*\text{OH}$ .

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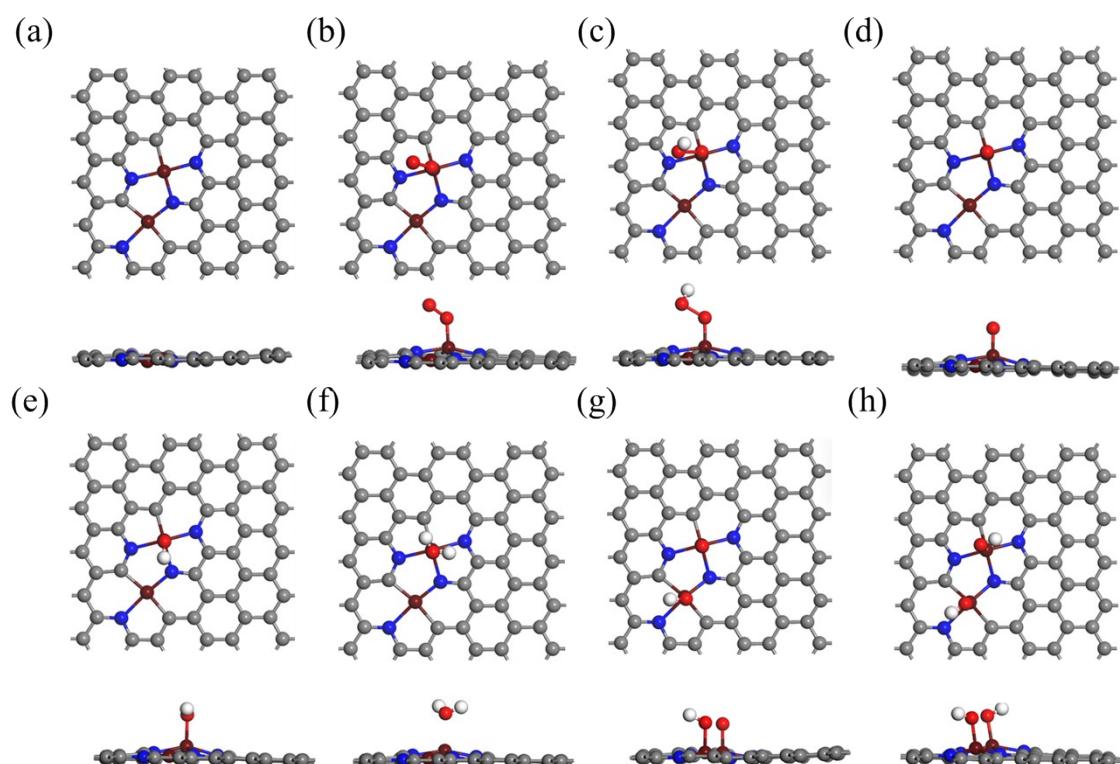


Fig. S13 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-}4\text{b}\text{-S2}/\text{G}$  catalytic surface.  
(a)\*, (b)  $^*\text{O}_2$ , (c)  $^*\text{OOH}$ , (d)  $^*\text{O}$ , (e)  $^*\text{OH}$ , (f)  $\text{H}_2\text{O}^*$ , (g)  $^*\text{O}^+*\text{OH}$ , (h)  $^*\text{OH}^+*\text{OH}$ .

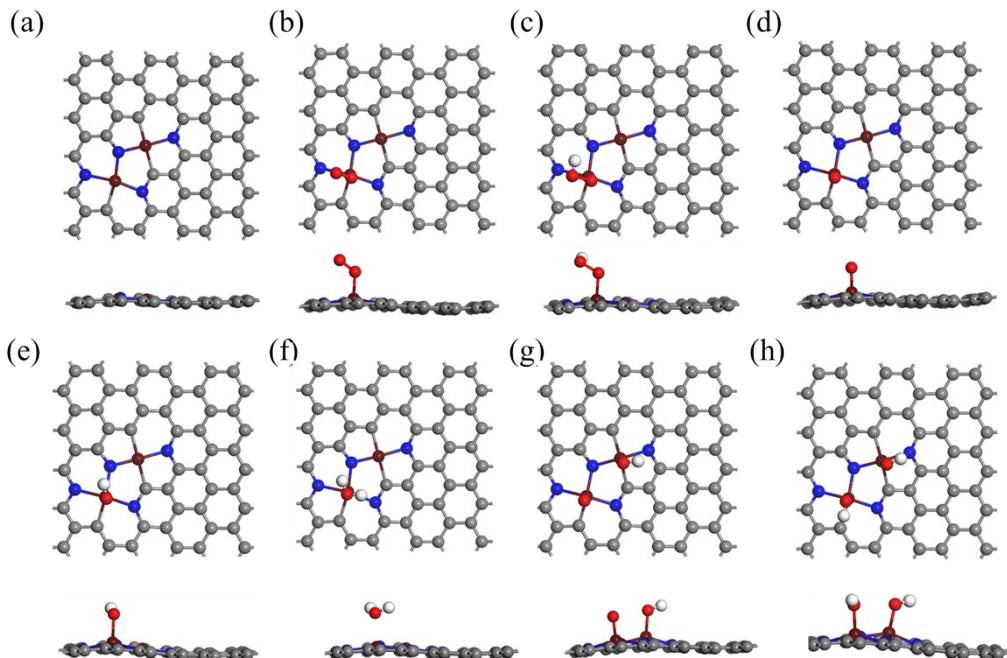


Fig. S14 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-5-S1/G}$  catalytic surface.  
(a)\*, (b)  $\text{O}_2^*$ , (c)  $\text{OOH}^*$ , (d)  $\text{O}^*$ , (e)  $\text{OH}^*$ , (f) $\text{H}_2\text{O}^*$ , (g)  $\text{O}^+\text{OH}^*$ , (h)  $\text{OH}^+\text{O}^*$ .

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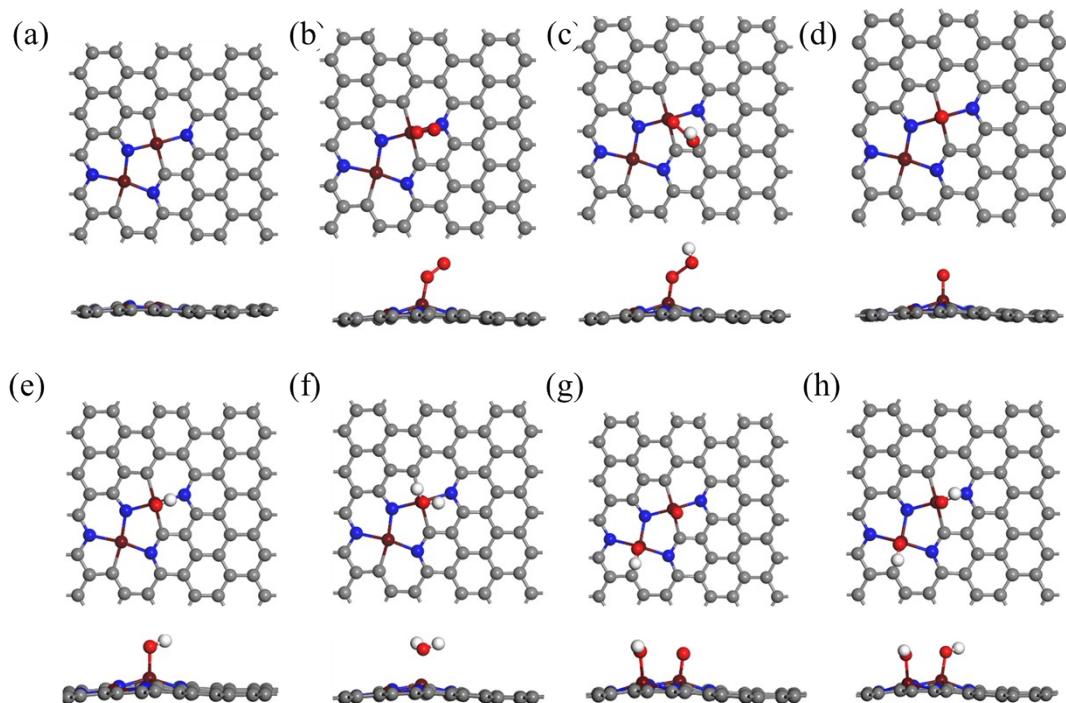


Fig. S15 Optimal structures of the oxygen-containing intermediates adsorbed on  $\text{Fe}_2\text{N}_4\text{-5-S2/G}$  catalytic surface.  
(a)\*, (b)  $\text{O}_2^*$ , (c)  $\text{OOH}^*$ , (d)  $\text{O}^*$ , (e)  $\text{OH}^*$ , (f) $\text{H}_2\text{O}^*$ , (g)  $\text{O}^+\text{OH}^*$ , (h)  $\text{OH}^+\text{O}^*$ .

Table S1. Bond length (Å) of the oxygen-containing intermediates on these catalysts

Active site	O <sub>2</sub>		OOH		O	OH	H <sub>2</sub> O	OH+OH			O+OH		
	M-O	O-O	M-O	O-O	M-O	M-O	M-O	O-M	O-O	M-O	M-O	M-O	O-O
FeN <sub>2</sub>	1.691	1.297	1.631	1.760	1.616	1.780	2.001	2.043	1.851	2.570	1.950	1.829	2.546
1	1.818	1.295	1.859	1.442	1.668	1.863	2.097	1.883	1.843	2.745	1.671	1.848	3.100
2	1.915	1.364	1.848	1.607	1.805	1.896	2.056	1.786	1.785	2.989	1.619	1.782	2.983
3	1.818	1.408	1.771	1.920	1.794	1.949	2.238	1.794	1.790	2.797	1.618	1.784	2.778
4a-S1	1.768	1.307	1.748	1.505	1.636	1.805	2.059	1.781	1.801	3.288	1.620	1.796	3.402
4a-S2	1.743	1.311	1.745	1.513	1.637	1.806	2.043	1.781	1.801	3.288	1.780	1.625	3.369
4b-S1	1.306	1.739	1.715	1.520	1.620	1.787	2.229	1.778	1.791	3.460	1.612	1.788	3.433
4b-S2	1.728	1.305	1.700	1.560	1.630	1.831	2.172	1.778	1.791	3.460	1.622	1.804	3.292
5a-S1	1.787	1.297	1.811	1.463	1.641	1.868	2.267	1.855	1.826	3.484	1.639	1.815	3.438
5a-S2	1.705	1.300	1.629	1.830	1.615	1.840	2.173	1.855	1.826	3.484	1.604	1.848	3.383

Table S2. Adsorption energies (eV) of the oxygen-containing intermediates on these catalysts

Active site	*O <sub>2</sub>	*OOH	*O	*OH	*H <sub>2</sub> O	Per co-ad *OH
FeN <sub>2</sub>	-1.43	-1.95	-5.00	-3.01	0.03	-2.44
1	-0.76	-1.33	-4.19	-2.48	-0.02	-2.56
2	-2.15	-2.56	-6.41	-3.61	-0.18	-3.02
3	-1.93	-2.57	-6.73	-3.79	-0.93	-3.06
4a-S1	-1.10	-1.45	-4.33	-2.60	0.20	-2.94
4a-S2	-1.40	-2.02	-4.58	-2.63	-0.36	-2.94
4b-S1	-0.98	-1.48	-4.50	-2.20	-0.10	-2.62
4b-S2	-1.16	-1.77	-4.78	-2.92	-0.29	-2.62
5a-S1	-0.71	-1.18	-4.19	-2.49	-0.16	-2.68
5a-S2	-1.14	-1.64	-4.87	-2.90	-0.26	-2.68